MATERIAL SAFETY DATA SHEET

Product Identity:

ARZOL SILVER NITRATE APPLICATORS

1.0 CHEMICAL PRODUCT AND COMPANY IDENTIFICATION

Product: ARZOL SILVER NITRATE APPLICATORS
Manufacturer's Name: Arzol Chemical Company
Address: 208 Benton Road, P.O. Box 91
North Haverhill, NH 03774
Date Issued: 11-19-93
Date Revised: Not yet revised

CHEMTREC CHEMICAL TRANSPORTATION EMERGENCY TELEPHONE NO.: 800-424-9300; D.C.: 202-489-7518
NATIONAL RESPONSE CENTER TELEPHONE NO. 800-424-0302; D.C. 202-426-2575

2.0 COMPOSITION/INFORMATION ON INGREDIENTS

<table>
<thead>
<tr>
<th>Ingredient</th>
<th>CAS Number</th>
<th>% Weight</th>
<th>OSHA Airborne Particulate Permissible Exposure Limit (PEL, TWA1)</th>
<th>ACGIH Airborne Particulate Threshold Limit Value (TLV, TWA1)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Silver nitrate</td>
<td>7761-88-8</td>
<td>75</td>
<td>0.01 mg/m³, as Ag metal and soluble compounds</td>
<td>0.1 mg/m³, as Ag metal</td>
</tr>
<tr>
<td>Potassium nitrate</td>
<td>7757-79-1</td>
<td>25</td>
<td>not established6</td>
<td>not established6</td>
</tr>
</tbody>
</table>

Note: The chemicals above are impregnated into the tip of a wooden applicator. The weight percentages indicated above represent the relative proportions of the active ingredients and do not take into account the weight of the applicator.

1 TWA - Values given are 8-hour time-weighted averages, unless otherwise specified.
2 Denoted ingredient is a SARA Title III, Section 313 listed toxic chemical (silver compounds).
3 Not established - Substance not assigned a specific PEL or TLV. Substance regulated by OSHA as particulates not otherwise regulated (PNOR, PELs - 15 mg/m³ total dust, 5 mg/m³ respirable fraction) and by ACGIH as particulates not otherwise classified (PNOC, TLV - 10 mg/m³, total dust containing no asbestos and less than 1% crystalline silica) and is considered a nuisance dust.
3.0 HEALTH HAZARDS IDENTIFICATION

EMERGENCY OVERVIEW: Silver nitrate/potassium nitrate solid impregnated on tip of wooden applicator is a strong irritant to skin and tissue. Toxic if ingested.

Primary Route(s) of Entry (for product): Inhalation: No Skin: Yes Ingestion: No Other: No

POTENTIAL HEALTH EFFECTS:

Note: Since the product has not been tested as a whole, the health effects below are based on the health effects of individual ingredients which are in significant concentrations in product. When appropriate, health effects of the individual ingredients are given in order to provide adequate warning to persons using the silver nitrate applicators. Silver nitrate applicators, in their final form for use, are not believed to pose high risk to the user, due to the small amount active ingredients on the tip of the applicator, and the highly remote likelihood of inadvertent or accidental exposures to toxic concentrations.

Acute Effects of Overexposure:

Eye contact: Contact with silver nitrate/potassium nitrate solid impregnated on tip of wooden applicator may cause irritation, the degree of which depends on the concentration and period of contact. Symptoms may include burning, tearing, and redness.

Skin contact: Contact with silver nitrate/potassium nitrate solid impregnated on tip of wooden applicator may cause irritation, the degree of which depends on the concentration and period of contact. Symptoms may include redness and burning.

Inhalation: Inhalation of airborne silver nitrate particles may cause irritation of mucous membranes.

Ingestion: Poisonous. If swallowed, can cause severe gastroenteritis and can be fatal. Due to its causticity, large doses of ingested silver nitrate burning sensation in the throat, violent abdominal pain, vomiting, collapse, and death.

Chronic Effects of Overexposure:

It is reported in the literature that chronic introduction of significant amounts of silver compounds into the blood stream and subsequent deposition of the reduced silver in various tissues of the body may result in the production of a generalized permanent grayish pigmentation of the skin and mucous membranes - a condition known as argyria, with no constitutional symptoms and no physical disability. The introduction of fine particles of silver through breaks in the skin produces a local pigmentation at the site of the injury. Localized argyria of the skin is rare. It has been concluded that on the average, 3.8 grams of orally administered silver nitrate causes argyria. The inhalation of silver powder over long periods has been concluded to cause pulmonary changes.

Chronic exposure to potassium nitrate can cause anemia, nephritis and methemoglobinemia.

Carcinogenicity: NTP: No IARC Monographs: No OSHA: No

Medical Conditions Generally Aggravated by Exposure (to silver nitrate): Preexisting diseases of the lungs, skin, eyes, and other mucous membranes.

4.0 FIRST AID MEASURES

Inhalation: This is not a probable route of exposure due to the product form. If acute overexposure to product occurs, immediately remove victim from the adverse environment to fresh air. If breathing is difficult, administer oxygen. If breathing has stopped, give artificial respiration. Seek medical attention immediately.
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Eye Contact: Wash out eye with lukewarm water for at least fifteen (15) minutes. Seek prompt medical attention.

Skin Contact: Wash skin area thoroughly with soap and water. Remove contaminated clothing. Seek medical attention.

Ingestion: Give one (1) glass of milk or water. Seek immediate medical attention. Never give anything by mouth to an unconscious person.

5.0 FIRE FIGHTING MEASURES

Flash Point (Method Used): For product, not applicable

Flammable limits: Not applicable for product

LEL: Not applicable

UEL: Not applicable

Autoignition Temperature: Not applicable for product

General Hazard: The impregnated solid is an oxidizer. May release toxic or irritating vapors under fire conditions.

Fire Fighting Instructions: As appropriate for surrounding fire. It is not believed that the product would be a significant hinderance to extinguishing methods used for the surrounding fire, due to the small amount of impregnated chemical solid and product form.

Fire Fighting Equipment: Fire fighters and others who may be exposed to combustion products during fire should wear full protective clothing, including self-contained breathing apparatus (SCBA). Wear SCBA with a full face-piece, operated in the positive pressure mode when fighting fires.

Hazardous Combustion Products: Acid, irritating smoke, oxides of nitrogen, potassium oxide, and oxides of carbon.

NFPA Rating:

Health: 1
Flammability: 1
Reactivity: 0
Special: OX (oxidizer)

6.0 ACCIDENTAL RELEASE MEASURES

Steps to be Taken in Case Material is Released or Spilled: Not applicable for product in final form (solid silver nitrate/potassium nitrate impregnated on tip of wooden applicator). Dispose of spent applicators in accordance with applicable federal, state, and local regulations.

7.0 HANDLING AND STORAGE

Storage Temperature and Pressure: Ambient temperature and pressure are adequate.

General: Store product in a dark, dry location, away from organic or other readily oxidizable materials. Keep container closed when not in use. Do not use in eyes. Keep away from children.

8.0 EXPOSURE CONTROLS/PERSOAL PROTECTION

Engineering Controls: Local and/or general ventilation, as needed, to reduce employee exposures to below applicable OSHA PELs and ACGIH TLVs (see SECTION 2.0, COMPOSITION INFORMATION ON INGREDIENTS, for PELs and TLVs). Due to final product form and use, it is not believed that PELs or TLVs will be exceeded.

Respiratory Protection: Use an appropriate NIOSH-approved respirator if airborne contaminant concentrations exceed applicable OSHA PEL or ACGIH TEL (see SECTION 2.0, COMPOSITION INFORMATION ON INGREDIENTS, for PELs and TLVs), or other Industry standards or guidelines on exposure. If respiratory protection is required, all appropriate
requirements as set forth in 29 CFR 1910.134 must be met. A competent health professional should be consulted for respirator selection. Due to final product form and use, it is not believed that PELs or TLVs will be exceeded.

Protective Gloves: Latex, vinyl or rubber examination gloves in order to prevent unnecessary or accidental skin contact.

Eye Protection: Safety glasses to prevent accidental contact.

Other Protective Clothing or Equipment: No special clothing necessary.

### 9.0 PHYSICAL AND CHEMICAL PROPERTIES

- **Boiling Point:** Decomposes, for silver nitrate
- **Vapor Pressure (mm Hg):** Not applicable
- **Vapor Density (AIR = 1):** Not applicable
- **Melting Point:** 414°F (212°C), for silver nitrate
- **Freezing Point:** Not applicable
- **Specific Gravity (H₂O = 1):** > 1, for silver nitrate
- **Evaporation Rate:** Not applicable
- **Solubility in water:** Soluble, for silver nitrate

**Appearance and Odor:** Grayish solid impregnated on tip of wooden applicator, practically odorless.

### 10.0 STABILITY AND REACTIVITY

**Stability:** Product is stable at ambient temperature and pressure. Exposure of product to light may cause oxidation and discoloration of the impregnated applicator.

**Conditions to avoid:** Contact of product with easily oxidizable materials and other incompatible materials. Heat or high temperature may cause solid in tip of product to decompose, possibly releasing small amounts of toxic or irritating vapors.

**Incompatible Materials:** Easily oxidizable materials.

- **Silver nitrate** is incompatible with alkalies, antimony salts, arsenites, bromides, carbonates, chlorides, lodides, thiocyanates, ferrous salts, hypophosphites, morphine salts, oils, creosote, phosphates, tannic acid, tartrates, vegetable decoctions and extracts; acetylene, acetylene + ammonium hydroxide, acetylene, ammonium hydroxide, arsenic, chlorine trifluoride, chlorosulfuric acid, ethanol, (magnesium powder + water), phosphorous, sulfur, charcoal, cuprous acetylide, magnesium, phosphine, phosphonium iodide, phosphorous laucyanate, and plastics.

- **Potassium nitrate** is incompatible with antimony, antimony trisulfide, arsenic, arsenic disulfide, bismuth sulfide, boron, boron phosphate, calcium sulfide, charcoal, copper phosphate, fluozine, germanium, germanium sulfide, sodium acetate, sodium hypophosphite, sodium peroxide + dextrose, sulfur + arsenic trisulfide, titanium, titanium disulfide, trichloroethylene, zinc, zirconium.

**Hazardous Decomposition:** When heated to decomposition, will emit small amounts of toxic NO₂ fumes and potassium oxide.

**Hazardous Polymerization:** Will Not Occur

### 11.0 TOXICOLOGICAL INFORMATION

For silver nitrate:

- **LD₅₀ oral mouse:** 50 mg/kg
- **LD₅₀ unknown route, man:** 20 mg/kg
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Eyes: rabbit - 1 mg, severe irritation

For potassium nitrate:

LD50 oral rabbit - 3015 mg/kg

Other toxicity data exists in the literature.

12.0 ECOLOGICAL INFORMATION

No data was available.

13.0 DISPOSAL CONSIDERATIONS

Waste Disposal Method: Dispose of in accordance with applicable federal, state, and local regulations. Consult an expert on the disposal of spent or recovered material. A solid waste determination should be performed by a qualified professional.

14.0 TRANSPORT INFORMATION

DOT hazard classification not determined.

15.0 REGULATORY INFORMATION

EPA SARA Title III Hazard Categorization: Based on the components of the tip of the silver nitrate applicators, the product is categorized as an immediate (acute) health hazard and delayed (chronic) health hazard.

EPA SARA Title III Section 302 Extremely Hazardous Substances (EHSs): No ingredients in this product are listed as an EHS under Section 302 of SARA Title III.

16.0 OTHER INFORMATION

All information, recommendations, and suggestions appearing herein concerning the product are based upon data believed to be reliable. It is the user's responsibility to determine the safety, toxicity, and suitability for their own use of the product described herein, and to comply with all applicable regulations. Since the actual use by others is beyond the MSDS developer's control, no guarantee, expressed or implied is made by Comprehensive Safety Compliance, Inc. (CSC acting consultant) as to the effects of such use, the results to be obtained, or the safety and toxicity of the product, nor does CSC assume any liability arising out of use by others of the product referred to herein. This MSDS is not intended as a license to operate under, or recommendation to infringe on, any patents. Appropriate warnings and safe handling procedures should be provided to handlers and users.

<table>
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<tr>
<th>PREPARED BY: Comprehensive Safety Compliance, Inc. (acting consultant)</th>
<th>REVISION NO.: N/A</th>
<th>DATE: November 19, 1993</th>
</tr>
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<tbody>
<tr>
<td>MSDS CONTACT:</td>
<td>SUPERSEDES MSDS DATED:</td>
<td>N/A</td>
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