Material Safety Data Sheet

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	PRODUCT NAME Carbon Dioxide,	
	Refrigerated Liquid TELEPHONE (415) 977-6500 EMERGENCY RESPONSE INFORMATION ON PAGE 2	
CARDOX CORPORATION	TRADE NAME AND SYNONYMS	CAS NUMBER
DIVISION OF LIQUID AIR CORPORATION	See last page.	124-38-9
One California Plaza, Suite 350 2121 N. California Blvd.	CHEMICAL NAME AND SYNONYMS Carbon	
Walnut Creek, California 94596	Dioxide, Carbonic Anhydride	
ISSUE DATE OCTOBER 1, 1985 AND REVISIONS CORPORATE SAFETY DEPT.	FORMULA MOLECULAR WEIGHT CO2 44.01	Carbonate
	HEALTH HAZARD DATA (SEE	NOTE ON LAST PAGE)
TIME WEIGHTED AVERAGE EXPOSURE LIMIT	OOO Walan DDW . It's CTFI is pro-	pased to be changed from
	,000 Molar PPM. Its STEL is prop Molar PPM (ACGIH, 1984-85).	oosed to be changed from
SYMPTOMS OF EXPOSURE		
reducing the oxygen level i include headache, dizziness ringing in the ears. High paralysis of the breathing atmosphere will cause a 50%	espiration is dependent on the CO n air, CO2 can cause suffocation, , shortness of breath, muscular v concentrations produce a faint ac control centers of the nervous sy increase in breathing rate; 3%, and is dangerous for even a few	yeakness, drowsiness and cid taste and can cause ystem: 2% by volume in the a 100% rate increase; >4%
TOXICOLOGICAL PROPERTIES Carbon dioxide is the most powerful	cerebral vasodilator known. Inhaling large and death. Chronic, harmful effects are not	concentrations causes rapid circula-
Rat, inhalation LCLo 657,190 ppm fo	r 15 minutes.	
Rat (10 days preg.), inhalation TCLo	60,000 ppm, 24 hours teratogenic effects.	
Human, inhalation TCLo 2,000 ppm		and the second of the second o
Frostbite effects are a change in the	color of the skin to gray or white possibly f	ollowed by blistering.
Listed as Carcinogen Nation or Potential Carcinogen Progr	nal Toxicology Yes 🗌 I.A.R.C. ram No 🗵 Monograph	Yes ☐ OSHA Yes ☐ s No ⊠ No ⊠
EXPOSURE TO CARBON DIOXIDE. BREATHING APPARATUS.		PPED WITH SELF-CONTAINED
fresh air. Quick removal f persons should be moved to and supplemental oxygen. A use of positional drainage.	ons should be assisted to an unco rom the contaminated area is most an uncontaminated area, given mou ssure that vomited material does Medical assistance should be so	important. Unconscious ith-to-mouth resuscitation not obstruct the airway by bught immediately.
Frostbite: Flush affected cian should see the patient of the dermal surface or de	areas with lukewarm water. DO NO promptly if the cryogenic "burn' ep tissue freezing.	OT USE HOT WATER. A physi- 'has resulted in blistering

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HAZARDOUS MIXTURES OF OTHER LIQUIDS, SOLIDS, OR GASES

Forms carbonic acid in the presence of water. See REACTIVITY DATA Section.

PHYSICAL DATA

BOILING POINT	LIQUID DENSITY AT BOILING POINT @ 0°F
Sublimation point = -109.3°F (-78.5°C)	Liquid Density = $63.65 \text{ lb/ft}^3 (1020 \text{ kg/m}^3)$
VAPOR PRESSURE	GAS DENSITY AT 70°F 1 atm @ 70°F (21.1°C) =
$0.70^{\circ}F(21.1^{\circ}C) = 844.7 \text{ psia} (5824 \text{ kPa})$	$1.1144 \text{ 1b/ft}^3 (1.832 \text{ kg/m}^3)$
SOLUBILITY IN WATER @68°F (20°C) Bunsen	FREEZING POINT -69.83°F (-56.57°C) @ 75.1 psia
Coefficient = .8704	(518 kPa)
APPEARANCE AND ODOR	
Colorless, odorless gas.	· · · · · · · · · · · · · · · · · · ·

FIRE AND EXPLOSION HAZARD DATA

FLASH POINT (METHOD USED)	AUTO IGNITION TEMPERATURE	FLAMMABLE LIMITS % BY VOLUME	
N/A	N/A	N/A	
Nonflammable, inert gas.		Rectrical classification Nonhazardous.	
containers to prevent rupture. This material is non-combustible. It can be used as a fire extinguishing agent primarily for its smothering effect (reduction of oxygen concentration to the point where the immediate atmosphere cannot support combustion.			
unusual fire and explosion Hazards It is not effective for use on fires involving chemicals that have their own oxygen supply (i.e., cellulose nitrate); or on fires involving reactive metals (such as, potassium, sodium, magnesium, aluminum, titanium, and zirconium), or their hydrides as these materials can decompose carbon dioxide.			

REACTIVITY DATA

STABILITY Unstable		conditions to Avoid CO2 is stable under ordinary conditions of use and storage. It does not polymerize. It does cause violent
Stable	X	polymerization of acrylaldehyde or ethyleneimine. It decomposes to CO and O2 when heated above (continued on last page)
peroxide w	(Materials to svoid) ith aluminum o	An explosion can occur when CO2 contacts mixtures of sodium r magnesium. Reactive metals (continued on last page)
HAZARDOUS DECC	OMPOSITION PRODUCTS	(sometimes on rase page)
Carbon mon	oxide.	
HAZARDOUS POLY May Occur	MERIZATION	CONDITIONS TO AVOID
Will Not Occur	X	Ay not a resp. of the control of the

SPILL OR LEAK PROCEDURES

STEPS TO BE TAKEN IN CASE MATERIAL IS RELEASED OR SPILLED

Evacuate area of major spill or release of CO2. Notify safety personnel. Provide ventilation. Clean-up personnel need special training and protection against contact with very cold materials or excessive inhalation of gaseous CO2.

WASTE DISPOSAL METHOD

Allow gas to bleed off at a moderate rate or solid to sublime to a well ventilated area.