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**** MATERIAL SAFETY DATA SHEET ****

Hydrochloric Acid 0.01 to 2.0N 40067

**** SECTION 1 - CHEMICAL PRODUCT AND COMPANY IDENTIFICATION ****

MSDS Name: Hydrochloric Acid 0.01 to 2.0N

MSDS Name: Hydrochiofic Acid v.01 to 2-200
Catalog Numbers:
S70041-2, S71944, S74855, S74856, S80036, S80039, SA621, GILHYDCHLOR, MCC--030293, MCC--030294, NC9619313, NC9655533, NC9668809, S70041-3, S718255, S74856MF, S80037, SA431 500, SA431500, SA481, SA4820, SA484, SA48500, SA5041, SA6820, SA5048, SA5020, SA5020, SA5020, SA5048, SA504, SA501, SA5020, SA5020, SA50201C, SA504, SA52 20, SA52 500, SA5220, SA52500, SA541, SA5410, SA5410, SA544 20, SA544, SA5411, SA5410, SA544, SA601, SA62-1, SA814, XX41704L, XX4200LI

Chlorohydric acid; Hydrogen chloride; Muriatic acid; Spirits of salt;

Hydrochloride Company Identification: Fisher Scientific

Fine Scientific Transport Lane 1 Reagent Lane 1 Rea

**** SECTION 2 - COMPOSITION, INFORMATION ON INGREDIENTS ****

CAS#	Chemical Name	%	EINECS#	
7647-01-0	Hydrochloric acid	0.03-6.1	231-595-7	
7732-18-5	Water	92.3-99.	231-791-2	h

Hazard Symbols: C

**** SECTION 3 - HAZARDS IDENTIFICATION ****

EMERGENCY OVERVIEW

Appearance: colorless to slight yellow.
Danger! Corrosive. Mutagen. May cause fetal effects based upon animal studies. Causes eye and skin burns. May cause severe respiratory tract irritation with possible burns. May cause severe digestive tract irritation with possible burns. Possible sensitizer. Target Organs: Respiratory system, eyes, skin, circulatory system.

Potential Health Effects

Eve:

May cause irreversible eye injury. Vapor or mist may cause irritation and severe burns. Contact with liquid is corrosive to the eves and causes severe burns. May cause painful sensitization to

May cause skin sensitization, an allergic reaction, which becomes evident upon re-exposure to this material. Contact with liquid is corrosive and causes severe burns and ulceration.

Ingestion:
May cause circulatory system failure. Causes severe digestive tract burns with abdominal pain, vomiting, and possible death. May cause corrosion and permanent tissue destruction of the esophagus and digestive tract.

May cause severe irritation of the respiratory tract with sore throat, coughing, shortness of breath and delayed lung edema. Causes chemical burns to the respiratory tract. Exposure to the mist and vapor may erode exposed teeth. Causes corrosive action on the mucous

Prolonged or repeated skin contact may cause dermatitis. Repeated exposure may cause erosion of teeth. May cause fetal effects. Laboratory experiments have resulted in mutagenic effects. Prolonged exposure may cause conjunctivitis, photosensitization, and possible blindness.

**** SECTION 4 - FIRST AID MEASURES ****

Get medical aid immediately. Do NOT allow victim to rub or keep eyes closed. Extensive irrigation is required (at least 30 minutes). SPEEDY ACTION IS CRITICAL!

ti Get medical aid immediately. Immediately flush skin with plenty of soap and water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. Destroy contaminated shoes Ingestion:

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Do NOT induce vomiting. If victim is conscious and alert, give 2-4 cupfuls of milk or water. Never give anything by mouth to an unconscious person. Get medical aid immediately. Give milk of magnesia.

nation:
Get medical aid immediately. Remove from exposure to fresh air immediately. If breathing is difficult, give oxygen. DO NOT use mouth-to-mouth respiration. If breathing has ceased apply artificial respiration using oxygen and a suitable mechanical device such as a bag and a mask.

Notes to Physician: Do Not use sodium bicarbonate in an attempt to neutralize the acid.

Antidote:
Do Not use oils or ointments in eye.

**** SECTION 5 - FIRE FIGHTING MEASURES ****

General Information:

neral Information:
As in any fire, wear a self-contained breathing apparatus in pressure-demand, MSHA/NIOSH (approved or equivalent), and full protective gear. Water runoff can cause environmental damage. Dike and collect water used to fight fire. During a fire, irritating and highly toxic gases may be generated by thermal decomposition or combustion. Not flammable, but reacts with most metals to form flammable hydrogen gas. Use water spray to keep fire-exposed containers cool. Vapors may be heavier than air. They can spread along the ground and collect in low or confined areas. Reaction with water may generate much heat which will increase the concentration. water may generate much heat which will increase the concentration of fumes in the air. Containers may explode when heated. Extinguishing Media:

For large fires, use water spray, fog, or alcohol-resistant foam.
Substance is nonflammable; use agent most appropriate to extinguish surrounding fire. Do NOT get water inside containers. Do NOT use straight streams of water. Most foams will react with the material straight streams of water, most loans win least with the material and release corrosive/toxic gases. Cool containers with flooding quantities of water until well after fire is out. For small fires, use carbon dioxide (expect for Cyanides), dry chemical, dry sand, and alcohol-resistant foam.

**** SECTION 6 - ACCIDENTAL RELEASE MEASURES ****

General Information: Use proper personal protective equipment as indicated in Section 8

Spills/Leaks: ills/Leaks:
Large spills may be neutralized with dilute alkaline solutions of soda ash, or lime. Avoid runoff into storm sewers and ditches which lead to waterways. Clean up spills immediately, observing precautions in the Protective Equipment section. Remove all sources of ignition. Provide ventilation. Do not get water inside containers. A vapor suppressing foam may be used to reduce vapors. Cover with dry earth, dry sand, or other non-combustible material followed with plastic sheet to minimize spreading and contact with water.

**** SECTION 7 - HANDLING and STORAGE ****

Idling:
Wash thoroughly after handling. Remove contaminated clothing and
wash before reuse. Use only in a well ventilated area. Contents may
develop pressure upon prolonged storage. Do not breathe dust, vapor,
mist, or gas. Do not get in eyes, on skin, or on clothing. Keep
container tightly closed. Do not ingest or inhale. Discard
contaminated shoes. Use caution when opening. Keep from contact with moist air and steam

rage:

Do not store in direct sunlight. Keep container closed when not in use. Store in a tightly closed container. Store in a cool, dry, well-ventilated area away from incompatible substances. Corrosives area. Do not store in metal containers. Do not store near flammable or oxidizing substances (especially nitric acid or chlorates)

**** SECTION 8 - EXPOSURE CONTROLS, PERSONAL PROTECTION ****

Engineering Controls:

ineering Controls.

Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use adequate general or local exhaust ventilation to keep airborne concentrations below the permissible exposure limits.

	Exposure L	imits			+
Chemical Name	ACGIH	,	NIOSH	OSHA - Final PELs	
Hydrochloric acid	C 5 ppm; C 7.5 mg/m3	50 p	pm IDLH n	C 5 ppm; C 7 ng/m3	
Water	none listed	none list	ed	none listed	+

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          OSHA Vacated PELs:
                 Hydrochloric acid:
                      No OSHA Vacated PELs are listed for this chemical
                      No OSHA Vacated PELs are listed for this chemical
         Personal Protective Equipment
                             Eyes:
                                      Wear appropriate protective eyeglasses or chemical safety goggles as described by OSHA's eye and face protection regulations in 29 CFR 1910.133 or European
                                       Wear neoprene or polyvinyl chloride gloves to prevent
                                       exposure
                      Clothing:
Wear appropriate protective clothing to prevent skin
                                       exposure
                  Respirators
                                       A respiratory protection program that meets OSHA's 29
CFR :1910.134 and ANSI 288.2 requirements or European
Standard EN 149 must be followed whenever workplace
                                       conditions warrant a respirator's use.
                       **** SECTION 9 - PHYSICAL AND CHEMICAL PROPERTIES ****
  Physical State:
                                                  Clear liquid
                                                colorless to slight yellow
strong, pungent
2.02(0.01N Solution)
5.7 mm Hg @ 0 deg C
  Appearance:
Odor:
  pH:
Vapor Pressure:
 Vapor Density:
Evaporation Rate:
                                                    1.26
1(N-butyl acetate = 1)
Viscosity: Not available.
Boiling Point: Not available.
Boiling Point: 81.5-110 deg C @ 760 mmHg
Freezing/Melting Point: -74 deg C
Autoignition Temperature: Not applicable.
Flash Point: Not applicable.
NFPA Rating: Not published
Explosion Limits | Not published
 RPPA Rating:

Explosion Limits, Lower:

Upper:

Not available.

Decomposition Temperature: Not available.

Solubility:

Miscible.

Specific Gravity/Density:

1.0 - 1.2
  Molecular Formula:
Molecular Weight:
                                                     36.46
                              **** SECTION 10 - STABILITY AND REACTIVITY ****
           Chemical Stability:
                  Stable under normal temperatures and pressures.
           Conditions to Avoid:
                   Mechanical shock, incompatible materials, contact with water, metals,
           excess heat, bases.
Incompatibilities with Other Materials:
                 omparionnus with Other Materials, Bases, acetic anhydride, alkali metals, aluminum, amines, copper, copper alloys, fluorine, iron, sodium hydroxide, steel, sulfuric acid, vinyl acetate, zinc, potassium permanganate, cesium acetylene carbide, rubidium carbide, sodium,
          carbide, rubidium acetylene carbide, rubidium carbide, sodium, chlorosulfonic acid, oleum, carbonates, perchloric acid, calcium phosphide, metal oxides, acetates, cesium carbide, beta-propiolactone, ethyleneimine, propylene oxide, lithium silicides, alcohols + hydrogen cyanide, 2-aminoethanol, ammonium hydroxide, calcium carbide, 1,1-difluoroethylene, ethylene diamine, magnesium boride, mercuric sulfate, silver perchlorate + carbon tetrachloride, uranium phosphide.

Hazardous Decomposition Products:

Hydrogen chloride, chlorine, carbon monoxide, carbon dioxide.
                   Hydrogen chloride, chlorine, carbon monoxide, carbon dioxide,
           hydrogen gas.
Hazardous Polymerization: Will not occur.
                            **** SECTION 11 - TOXICOLOGICAL INFORMATION ****
          CAS# 7647-01-0: MW4025000
CAS# 7732-18-5: ZC0110000
LD50/LC50:____
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CAS# 7647-01-0: Inhalation, mouse: LC50 =1108 ppm/1H; Inhalation, rat: LC50 =3124 ppm/1H; Oral, rabbit: LD50 = 900 mg/kg. CAS# 7732-18-5: Oral, rat: LD50 = 90 mL/kg.

Carcinogenicity:

Teratogenicity:

Hydrochloric acid -IARC: Group 3 carcinogen

Not listed by ACGIH, IARC, NIOSH, NTP, or OSHA.

Epidemiology:

Experimental reproductive effects have been reported.

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            Embryo or Fetus: Stunted fetus, Inhalation, rat TCL0=450 mg/m3/1H Specific Developmental Abnormalities: homeostatis, ihl-rat TCL0=450
      mg/m3/1H (female 1 days pre-mating).
Reproductive Effects:
No information available.
      Neurotoxicity:
            No information available
     Mutagenicity:
Cytogenetic analysis: Hamster, lung = 30 mmol/L.; Cytogenetic analysis: Hamster, ovary = 8 mmol/L.
            No data available.
                      **** SECTION 12 - ECOLOGICAL INFORMATION ****
      Ecotoxicity:
            Not available
            Fish: Bluegill/Sunfish: 3.6 mg/L; 48 Hr; Lethal (unspecified)
Fish: Bluegill/Sunfish: LD50; 96 Hr; pH 3.0-3.5
      Environmental Fate:
            Rapidly hydrolyzes when exposed to water. Will exhibit extensive
            evaporation from soil surfaces. Upon transport through the soil, hydrochloric acid will dissolve some of the soil materials (especially those with carbonate bases) and the acid will neutralize
             to some degree.
      Physical/Chemical:
            Not available.
     Other:
Not available
                     **** SECTION 13 - DISPOSAL CONSIDERATIONS ****
Dispose of in a manner consistent with federal, state, and local regulations. RCRA P-Series: None listed.
RCRA U-Series: None listed
                      **** SECTION 14 - TRANSPORT INFORMATION ****
      US DOT
            Shipping Name: HYDROCHLORIC ACID
Hazard Class: 8
                 UN Number: UN1789
      ON NUMBER: UN1789
Packing Group: II
Canadian TDG
Shipping Name: HYDROCHLORIC ACID SOLUTION
Hazard Class: 8(9.2)
UN Number: UN1789
                      **** SECTION 15 - REGULATORY INFORMATION ****
 US FEDERAL
      TSCA
         CAS# 7647-01-0 is listed on the TSCA inventory.
CAS# 7732-18-5 is listed on the TSCA inventory.
Health & Safety Reporting List
None of the chemicals are on the Health & Safety Reporting List.
Chemical Test Rules
Nace of the chemicals in this product are under a Chemical Test
            None of the chemicals in this product are under a Chemical Test Rule.
          Section 12b
None of the chemicals are listed under TSCA Section 12b
          TSCA Significant New Use Rule

None of the chemicals in this material have a SNUR under TSCA.
         Section 302 (RQ)
CAS# 7647-01-0: final RQ = 5000 pounds (2270 kg)
Section 302 (TPQ)
CAS# 7647-01-0: TPQ = 500 pounds; RQ = 5000 pounds (does not meet toxicity criteria but because of high production volume and recognized toxicity is considered a chemical of concern)
SARA Codes
            CAS # 7647-01-0: acute.
          Section 313

This chemical is not at a high enough concentration to be reportable under Section 313.
             No chemicals are reportable under Section 313.
      Clean Air Act:

CAS# 7647-01-0 is listed as a hazardous air pollutant (HAP).

This material does not contain any Class 1 Ozone depletors.

This material does not contain any Class 2 Ozone depletors.
      Clean Water Act:
CAS# 7647-01-0 is listed as a Hazardous Substance under the CWA.
             None of the chemicals in this product are listed as Priority Pollutants under the CWA.
             None of the chemicals in this product are listed as Toxic Pollutants
             under the CWA.
             CAS# 7647-01-0 is considered highly hazardous by OSHA
      Hydrochloric acid can be found on the following state right to know
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PAGE: 5 DATE: 01/27/00 ACCT: 888235001 INDEX: D00269285 CAT NO: SA49 PO NBR: 00001 lists: California, New Jersey, Florida, Pennsylvania, Minnesota, lists: California, New Jersey, Florida, Femily, Land, Massachusetts.

Water is not present on state lists from CA, PA, MN, MA, FL, or NJ. California No Significant Risk Level:
None of the chemicals in this product are listed.

European/International Regulations
European Labeling in Accordance with EC Directives
Hazard Symbols: C
Risk Phrases:
R 34 Causes burns. \$ 26 In case of contact with eyes, rinse immediately with plenty of water and seek medical advice. S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible). WGK (Water Danger/Protection)
CAS# 7647-01-0: 1
CAS# 7732-18-5: No information available. Canada

CAS# 7647-01-0 is listed on Canada's DSL/NDSL List.

CAS# 7732-18-5 is listed on Canada's DSL/NDSL List.

This product has a WHMIS classification of E, D2A.

CAS# 7647-01-0 is not listed on Canada's Ingredient Disclosure List.

CAS# 7732-18-5 is not listed on Canada's Ingredient Disclosure List. CAS# /64/-UI-U is not listed on Canada's ingredient Disclosure List.

Exposure Limits

CAS# 7637-01-0: OEL-AUSTRALIA:TWA 5 ppm (7 mg/m3)
OEL-AUSTRIA:TWA 5 ppm (7 mg/m3)
OEL-BELGIUM:STEL 5 ppm (7 mg/m3)
OEL-DENMARK:STEL 5 ppm (7 mg/m3)
OEL-FINLAND:STEL 5 ppm (7 mg/m3)
OEL-HUNGARY:STEL 5 ppm (7 mg/m3)
OEL-HUNGARY:STEL 5 ppm (7 mg/m3)
OEL-HUNGARY:STEL 5 ppm (7 mg/m3)
OEL-JAPAN:STEL 5 ppm (7 mg/m3)
OEL-THE NETHERLANDS:TWA 5 ppm (7 mg/m3)
OEL-THE PHILIPPINES:TWA 5 ppm (7 mg/m3)
OEL-POLAND:TWA 5 mg/m3
OEL-PUSIA:STEL 5 ppm (5 mg/m3)
OEL-SWEDEN:STEL 5 ppm (6 mg/m3)
OEL-SWEDEN:STEL 5 ppm (8 mg/m3)
OEL-SWEDEN:STEL 5 ppm (8 mg/m3)
OEL-THAILAND:TWA 5 ppm (7 mg/m3) **** SECTION 16 - ADDITIONAL INFORMATION **** MSDS Creation Date: 4/14/1999 Revision #1 Date: 11/23/1999 The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no way shall the company be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential or exemplary damages, howsoever arising, even if the company has been advised of the possibility of such damages.